

STUDY ON THERAPEUTIC USES FOR CERTAIN CHEMICAL COMPOUNDS USING COMPUTATIONAL AND SPECTROSCOPIC METHODS

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Abstract:

In order to investigate the potential therapeutic applications of certain chemical substances, this research article applies computational and spectroscopic methodologies. Researchers may learn more about the pharmacological characteristics and molecular interactions of these chemicals by combining computer modeling with spectroscopic investigation, which may help them find novel medication candidates and improve current treatments. The relevance of these multidisciplinary methods in furthering drug discovery and development is emphasized in the study, which also describes the methodology used and provides case examples of successful implementations.

Introduction:

For healthcare to advance and patient outcomes to improve, new therapeutic agents must be discovered and developed. Researchers continuously investigate new chemical compounds with potential therapeutic uses in order to meet these objectives. Scientists may now get important insights into the molecular interactions and pharmacological characteristics of these substances thanks to the development of computational and spectroscopic approaches as potent tools in pharmaceutical research.

The interactions between chemical substances and their target biomolecules may be predicted and studied by researchers via the use of computational approaches like molecular modeling and simulations. Researchers may select

interesting candidates for further inquiry by visually screening enormous chemical libraries, which will save time and money throughout the drug development process. QSAR analysis, which quantifies the link between a compound's structural characteristics and its biological activities, also helps in comprehending this relationship.

When describing chemical molecules, spectroscopic methods are very important in combination with computational methods. The chemical structure, composition, and behavior of substances may be investigated by researchers using spectroscopy, which includes NMR, FTIR, UV-Vis, and mass spectrometry. Researchers can confirm the existence of certain functional groups, evaluate the purity of compounds, and track chemical processes using spectroscopic analysis, giving crucial experimental data to support computational predictions.

A synergistic approach to drug discovery is provided by the combination of computational and spectroscopic approaches. Experimental design may be influenced by computational predictions, and computational models may be validated and improved by spectroscopic data. This integration improves scientists' comprehension of how compounds

behave, enabling logical medication development and the optimization of already-existing therapeutics.

This study intends to investigate the utilization and advantages of computational and spectroscopic approaches in analyzing the therapeutic applications of certain chemical substances. It will provide examples of how these multidisciplinary methods have led to important discoveries and achievements. The study will also go through the possible effects of these techniques on improving drug development, solving unmet medical needs, and ultimately advancing medical knowledge and patient care.

This work intends to emphasize the usefulness of multidisciplinary methods and encourage additional research in the area by emphasizing the relevance of computational and spectroscopic tools in drug discovery. Researchers may open the door for the creation of novel and successful therapeutic interventions by using the power of these techniques, ushering in a new age of medical therapies with enhanced effectiveness, safety, and patient outcomes.

Computational Methods for Drug Discovery:

Computational methods play a crucial role in drug discovery, allowing researchers to predict and analyze the interactions between chemical compounds and target biomolecules. These methods leverage the power of computers and algorithms to simulate and model molecular interactions, providing valuable insights into the potential therapeutic uses of specific chemical compounds. Some of the key computational methods used in drug discovery include:

1. **Molecular Docking:** Molecular docking is a computational technique used to predict the binding modes and affinities of small molecules (ligands) to target biomolecules (receptors). By simulating the interaction between the ligand and the receptor's binding site, researchers can identify potential drug candidates and understand their binding preferences.

2. **Molecular Dynamics Simulations:** Molecular dynamics simulations involve modeling the movement and behavior of atoms and molecules over time. By simulating the dynamics of a system, researchers can observe how a ligand interacts with a receptor in a dynamic environment, providing insights into the stability and flexibility of the ligand-receptor complex.

3. **Quantitative Structure-Activity Relationship (QSAR) Analysis:** QSAR is a computational method used to correlate the structural features of chemical compounds with their biological activities. Through mathematical models, researchers can predict the biological activity of new compounds based on their chemical structure, guiding the selection of promising drug candidates.

4. **Pharmacophore Modeling:** Pharmacophore modeling identifies the essential features or chemical groups required for a ligand to interact with a receptor and exhibit biological activity. By identifying common pharmacophoric features among active compounds, researchers can design new molecules with similar pharmacophores for improved activity.

5. **Virtual Screening:** Virtual screening involves using computational methods to virtually screen large chemical libraries and identify potential drug candidates. By prioritizing compounds

based on their predicted binding affinity and selectivity, virtual screening expedites the identification of lead compounds for further experimental validation.

6. **Structure-Based Drug Design:** Structure-based drug design utilizes the three-dimensional structure of target biomolecules, such as proteins or enzymes, to design new compounds with optimized interactions. By incorporating structural information, researchers can tailor ligands to fit specific binding sites and enhance their therapeutic potential.

7. **Ligand-Based Drug Design:** Ligand-based drug design relies on known active ligands to design new compounds with similar structures and biological activities. Computational methods, such as similarity searches and molecular fingerprints, aid in identifying structurally related compounds for drug discovery.

These computational methods offer significant advantages in drug discovery, including the ability to screen vast chemical databases, understand molecular interactions at the atomic level, and prioritize potential drug candidates for experimental validation. By complementing experimental approaches, computational methods accelerate the drug discovery process, leading to the development of more effective and targeted therapeutic agents.

Spectroscopic Techniques in Drug Characterization:

Spectroscopic techniques are powerful tools used in drug characterization to analyze the chemical structure, composition, and behavior of chemical compounds. These methods provide valuable experimental data that complement computational predictions and aid in understanding the properties and interactions of potential drug candidates.

Some of the key spectroscopic techniques used in drug characterization include:

1. **Nuclear Magnetic Resonance (NMR) Spectroscopy:** NMR spectroscopy is a widely used technique for determining the three-dimensional structure of molecules in solution. It provides information about the connectivity and arrangement of atoms within a compound, allowing researchers to identify functional groups and assess compound purity. NMR can also be used to study ligand-receptor interactions, providing insights into the binding modes of drug candidates to target biomolecules.

2. **Fourier Transform Infrared Spectroscopy (FTIR):** FTIR spectroscopy measures the vibrational modes of molecules, providing information about their chemical bonds and functional groups. It is particularly useful in identifying specific chemical groups in a compound and studying molecular conformational changes. FTIR is commonly used in quality control to verify the chemical identity and composition of drug substances.

3. **Ultraviolet-Visible (UV-Vis) Spectroscopy:** UV-Vis spectroscopy measures the absorption of light in the ultraviolet and visible regions of the electromagnetic spectrum. It is often used to determine the presence and concentration of chromophores in chemical compounds. UV-Vis spectroscopy is valuable in studying the stability and degradation of drugs and monitoring enzymatic reactions.

4. **Mass Spectrometry (MS):** Mass spectrometry is a technique used to measure the mass-to-charge ratio of ions, providing information about the molecular weight and structural composition of compounds. MS is used for compound

identification, quantification, and studying fragmentation patterns. It is an essential tool in drug metabolism studies and the analysis of drug impurities.

5. **Circular Dichroism (CD)**

Spectroscopy: CD spectroscopy measures the differential absorption of left- and right-circularly polarized light by chiral molecules. It is used to determine the secondary structure of proteins and assess changes in protein conformation induced by ligand binding. CD spectroscopy is valuable in studying protein-ligand interactions and protein stability.

6. **Raman Spectroscopy:** Raman spectroscopy measures the inelastic scattering of light, providing information about molecular vibrations and crystal structures. It is useful in studying solid-state properties of drugs and characterizing polymorphs, which can affect a drug's solubility and bioavailability.

7. **Fluorescence Spectroscopy:**

Fluorescence spectroscopy measures the emission of light from fluorophores upon excitation with light of a specific wavelength. It is used to study protein-protein interactions, ligand-receptor binding, and enzyme kinetics. Fluorescence spectroscopy is particularly valuable in drug target validation and mechanism of action studies.

Spectroscopic techniques offer non-destructive and highly sensitive methods for drug characterization, providing critical information for drug development and quality control. By analyzing the chemical properties and interactions of chemical compounds, spectroscopy enhances our understanding of drug behavior and contributes to the rational design and optimization of therapeutic agents.

Integration of Computational and Spectroscopic Approaches:

The integration of computational and spectroscopic approaches in drug research is a powerful and synergistic strategy that leverages the strengths of both methods. By combining computational modeling with experimental spectroscopic data, researchers can gain a more comprehensive understanding of the behavior and interactions of chemical compounds, particularly those with potential therapeutic uses. This integration facilitates drug discovery, optimization, and validation processes in several ways:

1. **Validation of Computational Models:**

Spectroscopic data can be used to validate and refine the accuracy of computational models. Experimental spectroscopy provides direct evidence of the presence of specific functional groups, molecular conformations, and interactions that were predicted by computational methods. This validation enhances the reliability of computational predictions and increases researchers' confidence in the identified drug candidates.

2. **Characterization of Ligand-Receptor Interactions:**

Combining computational docking studies with spectroscopic techniques, such as NMR or CD spectroscopy, allows researchers to characterize ligand-receptor interactions in detail. This integration provides insights into the binding modes, orientation, and stability of drug candidates within the target binding site, helping to elucidate the molecular mechanisms of drug action.

3. **Structure Elucidation and Conformational Analysis:**

Spectroscopic methods, such as NMR and FTIR, offer valuable information on the three-dimensional structure and conformational changes of chemical compounds. By comparing experimental spectroscopic data with computational models,

researchers can refine the structural information, enhancing the accuracy of drug design and optimization.

4. Binding Affinity and Thermodynamics: The combination of computational and spectroscopic methods allows for the determination of binding affinities and thermodynamic parameters of ligand-receptor interactions. This information is essential for understanding the energetics of ligand binding and optimizing drug candidates for enhanced potency and selectivity.

5. Mechanism of Action Studies: Integrating spectroscopic data with computational simulations enables researchers to study the dynamic behavior of drug-target complexes over time. This information is crucial for investigating the mechanism of action and kinetics of drug binding, shedding light on how drug compounds exert their therapeutic effects.

6. Virtual Screening and Experimental Validation: Virtual screening using computational methods can efficiently prioritize potential drug candidates from large chemical databases. Spectroscopic techniques can then be used to experimentally validate the binding affinity and functional properties of the selected compounds, streamlining the process of lead compound identification.

7. Drug Formulation and Delivery: Spectroscopic analysis plays a crucial role in drug formulation and delivery studies. By using spectroscopy to investigate drug-excipient interactions and stability, researchers can optimize drug formulations for improved bioavailability and efficacy.

The integration of computational and spectroscopic approaches in drug research offers a powerful and synergistic approach to understanding the therapeutic potential

of chemical compounds. This combined approach enhances the reliability of computational predictions, provides valuable experimental data for validation, and offers a deeper understanding of ligand-receptor interactions and drug mechanisms of action. By leveraging the strengths of both methods, researchers can accelerate drug discovery, design more effective therapies, and contribute to the advancement of medical science and patient care.

Conclusion

In conclusion, the research paper highlights the importance of computational and spectroscopic methods in investigating the therapeutic uses of certain chemical compounds. These interdisciplinary approaches play a crucial role in drug discovery, optimization, and validation, contributing to the advancement of medical science and patient care. By leveraging computational modeling and spectroscopic analysis, researchers can gain valuable insights into the molecular interactions, pharmacological properties, and mechanisms of action of chemical compounds with potential medical applications.

Computational methods, such as molecular docking, molecular dynamics simulations, QSAR analysis, and pharmacophore modeling, enable researchers to predict and analyze the interactions between chemical compounds and target biomolecules. These predictions aid in virtual screening of chemical databases, identifying potential drug candidates for further experimental validation, and optimizing lead compounds for enhanced potency and selectivity.

On the other hand, spectroscopic techniques, including NMR, FTIR, UV-Vis, Mass Spectrometry, CD spectroscopy,

Raman spectroscopy, and fluorescence spectroscopy, provide experimental data that complement computational models. Spectroscopy allows for the characterization of chemical compounds, determination of molecular structures, validation of ligand-receptor interactions, and investigation of drug mechanisms of action. Moreover, it aids in drug formulation studies and quality control, ensuring the purity and stability of drug substances.

The integration of computational and spectroscopic approaches offers a synergistic strategy that enhances the reliability and accuracy of drug research. Spectroscopic data validate computational predictions, providing direct evidence of the presence of specific functional groups and molecular interactions. In turn, computational models guide experimental design and help interpret spectroscopic data, leading to a more comprehensive understanding of drug behavior.

REFERENCES

1. Abbanat, D.; Macielag, M.; Bush, K. "Novel antibacterial agents for the treatment of serious gram-positive infections", *Expert Opin. Investig. Drugs* 12, 2003, 79-399.
2. Abbas, S.Y.; El-Sharief, M.A.M.; Basyouni, W.M.; Fakhr, I.M.; El-Gammal, E.W. "Thiourea derivatives incorporating a hippuric acid moiety, synthesis and evaluation of antibacterial and antifungal activities", *Eur. J. Med. Chem.* 64, 2013, 111-120.
3. Abraham, R.J.; Mobli, M.; Smith, R.J. "1H chemical shifts in NMR: Part 19. Carbonyl anisotropies and steric effects in aromatic aldehydes and ketones", *Magn. Reson. Chem.* 41, 2003, 26-36.
4. Abramović, B.; Kler, S.; Sojic, D.; Lausevic, M.; Radovic, T.; Vione, D. "Photocatalytic degradation of metoprolol tartrate in suspensions of two TiO₂ based photocatalysts with different surface area, identification of intermediates and proposal of degradation pathways", *J. Hazard. Mater.* 198, 2011, 123-132.
5. Adant, C.; Dupuis, M.; Bredas, J.L. "Ab initio study of the nonlinear optical properties of urea : Electron correlation and dispersion effects", *Int. J. Quantum. Chem.* 56, 1995, 497-507.
6. Addla, D.; Jallapally, A.; Gurram, D.; Yogeewari, P.; Sriram, D.; Kantevari, S. "Design, synthesis and evaluation of 1,2,3-triazoleadamantylacetamide hybrids as potent inhibitors of Mycobacterium tuberculosis", *Bioorg. Med. Chem. Lett.* 24, 2014, 1974-1979.
7. Ahmad, S.; Mathew, S.; Verma, P.K. "Laser Raman and FT-infrared spectra of 3,5-dinitrobenzoic acid", *Indian J. Pure Appl. Phys.* 30, 1992, 764-765.
8. Ai, L.I.; Liu, J.Y. "Mechanism of OH-initiated atmospheric oxidation of E/Z-CF₃CF=CF₃: a quantum mechanical study", *J. Mol. Model.* 20, 2014, 1-10.
9. Ajaj, I.; Markovski, J.; Marković, J.; Jovanović, M.; Milčić, M.; Assaleh, F.; Marinković, A. "Solvent and structural effects in tautomeric 3-cyano-4- (substituted phenyl)-6-phenyl-2 (1H)-pyridones: experimental and quantum chemical study", *Structural Chemistry*, 25, 2014, 1257-1270.
10. Akbay, A.; Oren, I.; Temiz-Arpaci, O.; Aki-Sener, E.; Yalcin, I. "Synthesis and HIV-1 reverse transcriptase inhibitor activity of some 2,5,6-substituted benzoxazole, benzimidazole, benzothiazole and oxazolo (4,5-b) pyridine derivatives", *Arzneim. Forsch.* 53, 2003, 266-271.
11. Al-Abdullah, E.S.; Asiri, H.H.; El-Emam, A.; Ng, S.W. "3-(Adamantan-1-yl)-4-phenyl-1-[(4-phenylpiperazin-1-yl)methyl]-1H-1,2,4-triazole-5(4H)-thione", *Acta Cryst. E* 68, 2012, 345.
12. Al-Abdullah, E.S.; Asiri, H.H.; Lahsasni, S.; Habib, E.E.; Ibrahim, T.M.; El-Emam, A.A.; "Synthesis, antimicrobial, and anti-inflammatory activity, of novel S-substituted and N-substituted 5-(1-adamantyl)-1,2,4-triazole-3- thiols", *Drug Des. Dev. Ther.* 8, 2014, 505-518.
13. Al-Abdullah, E.S.; Sebastian, S.H.R.; Al-Wabli, R.I.; El-emam, A.A.; Panicker, C.Y.; Van Alsenoy, C. "Vibrational spectroscopic studies (FT-IR, FT-Raman) and quantum chemical calculations on 5-(adamantan-1-yl)- 3-[(4-fluoroanilino)methyl]-2,3-dihydro-1,3,4-oxadiazole-2-thione, a potential chemotherapeutic agent", *Spectrochim. Acta A* 133, 2014, 605- 618.
14. Al-Deeb, O.A.; Al-Omar, M.A.; El-Brollosy, N.R.; Habib, E.E.; Ibrahim, T.M.; El-

- Emam, A.A. "Synthesis, antimicrobial, and anti-inflammatory activities of novel 2-[3-(1-adamantyl)-4-substituted-5-thioxo-1,2,4-triazolin-1-yl]acetic acids, 2-[3-(1-adamantyl)-4-substituted-5-thioxo-1,2,4-triazolin-1-yl]propionic acids and related derivatives", *Arzneim.-Forsch./Drug Res.* 56, 2006, 40-47.
15. Algorithm for protein ligand docking", *PROTEINS: Struct. Funct. Genet.* 37, 1999, 228-241.
16. Alkan, C.; Tek, Y.; Kahraman, D. Preparation and characterization of a series of thiourea derivatives as phase change materials for thermal energy storage, *Turkish J. Chem.* 35, 2011, 769-777.
17. Almajan, G.L.; Barbuceanu, S.; Almajan, E.; Draghici, C.; Saramet, G.; "Synthesis, characterization and antibacterial activity of some triazoleMannich bases carrying diphenylsulfone moieties", *Eur. J. Med. Chem.* 44, 2009, 3083-3089.
18. Al-Omary, F.A.M.; Mary, Y.S.; Panicker, C.Y.; El-Emam, A.A.; Al-Swaidan, I.A.; Al-Saadi, A.A.; Van Alsenoy, C.; "Spectroscopic investigations, NBO, HOMO-LUMO, NLO analysis and molecular docking 5-(adamantan-1-yl)-3-anilinomethyl-2,3-dihydro-1,3,4-oxadiazole-2-thione, a potential bioactive agent", *J. Mol. Struct.* 1096, 2015, 1-14.
19. Alper-Hayta, S.; Arisoy, M.; Temiz-Arpaci, O.; Yildiz, I.; Aki, E.; Ozkan, S.; Kaynak, F. "Synthesis, antimicrobial activity, pharmacophore analysis of some new 2-(substitutedphenyl/benzyl)-5-[(2-benzofuryl)carboxamido] benzoxazoles", *Eur. J. Med. Chem.*, 43, 2008, 2568-2578.
20. Al-Wabli, R.I.; Resmi, K.S.; Mary, Y.S.; Panicker, C.Y.; Attia, M.A.; Al-Emam, A.A.; Van Alsenoy, C. "Vibrational spectroscopic studies, Fukui functions, HOMO-LUMO, NLO, NBO analysis and molecular docking study of (E)-1-(1,3-benzodioxol-5-yl)-4,4-dimethylpent-1-en-3-one, a potential precursor to bioactive agents", *J. Mol. Struct.* 1123, 2016, 375- 383.
21. Andersson, T.; Broo, A.; Evertsson, E. "Prediction of drug candidates sensitivity toward autoxidation: computational estimation of C-H dissociation energies of carbon centered radicals", *J. Pharm. Sci.* 103, 2014, 1949-1955.
22. Armaković, S.; Armaković, S.J.; Abramović, B.F. "Theoretical investigation of loratadine reactivity in order to understand its degradation properties: DFT and MD study", *Journal of molecular modeling* 22, 2016, 240.
23. Armaković, S.; Armaković, S.J.; Koziel, S. "Optoelectronic properties of curved carbon systems," *Carbon*, 111, 2017, 371-379.
24. Armaković, S.; Armaković, S.J.; Šetrajčić, J.P.; Šetrajčić, I.J. "Active components of frequently used β -blockers from the aspect of computational study", *J. Mol. Model.* 18, 2012, 4491-4501.
25. Armaković, S.J.; Armaković, S.; Finčur, N.L.; Šibul, F.; Vione, D. Šetrajčić, J.P.; Abramović, B. Influence of electron acceptors on the kinetics of metoprolol photocatalytic degradation in TiO₂ suspension: A combined experimental and theoretical study, *RSC Advances* 5, 2015, 54589-54604.
26. Armaković, S.J.; Grujić-Brojčin, M.; Šćepanović, M.; Armaković, S.; Golubović, A.; Babić, B.; Abramović, B.F. "Efficiency of La-doped TiO₂ calcined at different temperatures in photocatalytic degradation of β -blockers", *Arabian Journal of Chemistry*, 2017.
27. Ataly, Y.; Avci, D.; Basoglu, A. Linear and nonlinear optical properties of some donor-acceptor oxadiazoles by ab initio Hartree-Fock calculations, *Struct. Chem.* 19, 2008, 239-246.
28. Balzarini, J.; Orzeszko, B.; Mauri, J.K.; Orzeszko, A. "Synthesis and anti-HIV studies of 2-adamantyl-substituted thiazolidin-4-ones", *Eur. J. Med. Chem.* 42, 2007, 993-1003.
29. Banks, J.L.; Beard, H.S.; Cao, Y.; Cho, A.E.; Damm, W.; Farid, R.; Felts, A.K.; Halgren, T.A.; Mainz, D.T.; Maple, J.R.; Murphy, R.; Philipp, D.M.; Repasky, M.P.; Zhang, L.Y.; Berne, B.J.; Friesner, R.A.; Galicchio, E.; Levy, R.M. "Integrated modeling program applied chemical theory (IMPACT)", *J. Comput. Chem.* 26, 2005, 1752-1780.
30. Becke, A.D. Density functional thermochemistry. III. The role of exact exchange, *J. Chem. Phys.* 98, 1993, 5648-5652.
31. Becke, A.D. "Density functional exchange energy approximation with correct asymptotic behavior", *Phys. Rev. A* 38, 1988, 3098-3100
32. Benzon, K.B.; Varghese, H.T.; Panicker, C.Y.; Pradhan, K.; Tiwary, B.K.; Nanda, A.K.; Van Alsenoy, C. "Spectroscopic investigation (FT-IR and FT-Raman), vibrational assignments, HOMO-LUMO, NBO, MEP analysis and molecular docking study of 2-(4-hydroxyphenyl)4,5-dimethyl-1H-imidazole 3-oxide", *Spectrochim. Acta* 146,

2015, 307-322.

33. Berendsen, H.J.; Postma, J.P.; Van Gunsteren, W.F.; Hermans, J. "Interaction models for water in relation to protein hydration, in Intermolecular forces." Springer. p., 1981, 331-342.
34. Bhagyasree, J.B.; Samuel, J.; Varghese, H.T.; Panicker, C.Y.; Arisoy, M.; Temiz-Arpaci, O. "Synthesis, FT-IR investigation and computational study of 5-[(4-bromophenyl)acetamido]-2-(4-tert-butylphenyl)benzoxazole.", *Spectrochim. Acta* 115, 2013, 79-91.
35. Bhagyasree, J.B.; Varghese, H.T.; Panicker, C.Y.; Samuel, J.; Van Alsenoy, C.; Bolelli, K.; Yildiz, I.; Aki, E. "Vibrational spectroscopic (FT- IR, FT-Raman, ¹H NMR and UV) investigations and computational study of 5-nitro-2-(4-nitrobenzyl)benzoxazole.", *Spectrochim. Acta* 102, 2013, 99-113.
36. Bhowruth, V.; Brown, A.K.; Reynolds, R.C.; Coxon, G.D.; Mackay, S.P.; Minnikin, D.E.; Besra, G.S. "Symmetrical and unsymmetrical analogues of isoxyl: active agents against *Mycobacterium tuberculosis*", *Bioorg. Med. Chem. Lett.* 16, 2006, 4743-4747.
37. Bielenica, A.; Kedzierska, E.; Fidecka, S.; Maluszynska, H.; Mirosław, B.; Koziol, A.E.; Stefanska, J.; Madeddu, S.; Giliberti, G.; Sanna, G.; Struga, M. "Synthesis, antimicrobial and pharmacological evaluation of thiourea derivatives of 4H-1,2,4-triazole", *Lett. Drug Des. Discov.* 12, 2015, 263-276.
38. Bielenica, A.; Kędzierska, E.; Koliński, M.; Kmieciak, S.; Koliński, A.; Fiorino, F.; Severino, B.; Magli, E.; Corvino, A.; Rossi, I.; Massarelli, P.; Koziol, A.E.; Sawczenko, A.; Struga, M. "5-HT₂ receptor affinity, docking studies and pharmacological evaluation of a series of 1,3-disubstituted thiourea derivatives.", *Eur. J. Med. Chem.* 116, 2016, 173-186.
39. Bielenica, A.; Stefańska, J.; Stępień, K.; Napiórkowska, A.; Augustynowicz-Kopeć, E.; Sanna, G.; Madeddu, S.; Boi, S.; Giliberti, G.; Wrzosek, M.; Struga, M. "Synthesis, cytotoxicity and antimicrobial activity of thiourea derivatives incorporating 3-(trifluoromethyl)phenyl moiety", *Eur. J. Med. Chem.* 101, 2015, 111-125.
40. Bielenica, A.; Stępień, K.; Napiórkowska, A.; Augustynowicz-Kopeć, E.; Krukowski, S.; Włodarczyk, M.; Struga, M. "Synthesis and antimicrobial activity of 4-chloro-3-nitrophenylthiourea derivatives targeting bacterial type II topoisomerases", *Chem. Biol. Drug Des.* 87, 2016, 905-917.